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# **ORIGINAL ARTICLE**

## Three new secoiridoid glycoside dimers from Swertia mileensis

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Three new secoiridoid glycoside dimers named swerilactosides A-C (1–3) were isolated from *Swertia mileensis*. Their structures were elucidated based on extensive spectral analyses (1D and 2D NMR, MS, and IR spectroscopic means).

Keywords: secoiridoid glycoside dimers; swerilactosides A-C; Swertia mileensis; Gentianaceae

## 1. Introduction

The family Gentianaceae, annual or perennial herbs, contains about 80 genera and 700 species, of which 22 genera and 427 species are distributed in China [1]. Many species mainly belonging to the Gentiana and Swertia genus are used as traditional Chinese herbs to treat hepatitis, cholecystitis, and digestive system disease [2]. Previous investigation reveals that secoiridoid glycosides, xanthones, flavones, and triterpenoids are the main constituents of Gentianaceae plants [3]. In 1958, Fu and Sun [4] reported the isolation of three alkaloids from Gentiana macrophylla (namely 'Qin-Jiao' in Chinese), one of which was identified as gentianine. Afterwards, Prof. Liang et al. [5,6] first applied NMR and IR spectral analyses, together with chemical methods, to determine the structures of gentinidine and gentianal. Later, Govindachari et al. [7] proved gentianine to be an artificial product during the extraction with  $NH_3 \cdot H_2O$ .

Swertia mileensis (=Swertia leducii), well known as 'Qing-Ye-Dan' in Chinese, belongs to the Swertia genus, the second largest genus next to Gentiana of the family Gentianaceae [8]. As a traditional Chinese medicine (TCM), it has long been used to treat viral hepatitis in the Yi and Ha-Ni nationality regions, Mile and Kaiyuan Counties, Yunnan Province. In the 1970s, a large amount of phytochemical and pharmacological investigations on S. mileensis was carried out, which promoted it to be documented in Chinese Pharmacopoeia (1977-2010 editions) as a new TCM source [9]. Presently, its significantly curative effect on acute viral hepatitis has resulted in wide clinical applications [10–13].

In order to clarify the active components [14], our previous bioassay-guided fractionation has led to the isolation of four types of novel iridoid lactones:

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swerilactones A and B (C18 skeleton) [15], swerilactones C and D (C20 skeleton) [16], swerilactones E and F (lactones with naphthyl rings), and swerilactone G (a secoiridoid aglycone dimer) [17], with anti-HBV activity in vitro, and subsequently, the other three unusual secoiridoid glycoside dimers (two molecules of secoiridoids connected by a molecule of the glycosyl group) were obtained from this plant. Generally, the Swertia genus is rich in secoiridoid glycosides; however, the secoiridoid glycoside dimers were seldom reported [3,18]. Herein, we describe the isolation and structural elucidation of swerilactosides A-C based on extensive spectroscopic analyses (Figure 1).

## 2. Results and discussion

Swerilactoside A (1) had a molecular formula of  $C_{25}H_{32}O_{13}$  by positive HR-ESI-MS at m/z 563.1728 [M + Na]<sup>+</sup>. The IR spectrum showed the absorption bands of OH (3423 cm<sup>-1</sup>), C-O (1698 cm<sup>-1</sup>), double bond (1620 cm<sup>-1</sup>), and glycosyl group (1082, 1027, and 1005 cm<sup>-1</sup>).

The <sup>1</sup>H and <sup>13</sup>C NMR spectra of compound **1** displayed 25 carbon signals due to 6 quaternary carbons, 11 methines, 7 methylenes, and 1 methyl group, of which two lactone carbonyl carbons, three double bonds, and one glucosyl group were revealed (Table 1).

Detailed analyses of its NMR spectra suggested a swertiamarin fragment (1a), which was also supported by the  ${}^{1}\text{H}{-}{}^{1}\text{H}$  COSY, HMBC, and ROESY spectra. Compared to the known swertiamarin [19], the C-3' in compound 1 was shifted significantly downfield from  $\delta_{\rm C}$  77.7 (d) to 86.3 (d); contrarily, C-4' was shifted slightly upfield from  $\delta_{\rm C}$  71.4 (d) to 69.5 (d), which suggested that another partial structure was linked to C-3' by the glycosidic linkage in compound 1.

In addition to the swertiamarin fragment (1a), the nine residual carbons were ascribed to one lactone carbonyl carbon  $[\delta_{\rm C} \ 166.5 \ ({\rm s, C-10}'')],$  one tetra-substituted double bond [ $\delta_{\rm C}$  158.3 (s, C-5") and 123.2 (s, C-4")], two oxygenated methines [ $\delta_{\rm C}$ 95.8 (d, C-3", dioxygenated one) and 63.4 (d, C-1")], three methylenes [ $\delta_{\rm C}$  67.4 (t, C-7'', oxygenated one), 37.3 (t, C-8''), and 29.0 (t, C-6")], and one methyl group [ $\delta_{\rm C}$ 20.6 (q, C-9'')], which indicated a secoiridoid aglycone-like fragment. In the HMBC spectrum, the correlations of H-7" with C-5" and C-10", H-6" with C-4" and C-8", H-8" with C-4" and C-9", and H-3'' with C-1'' and C-10'', together with the <sup>1</sup>H – <sup>1</sup>H COSY correlations of H-7"/H-6" and H-8"/H-1"/H-9", led to the construction of the partial fragment 1b (Figure 2).

The connection of C-3' and C-3" by an ether bond was determined by the HMBC correlations of H-3' with C-3" and H-3"



Figure 1. The structures of compounds 1-3.

No. 1 1 5.74, 4 7.63, 5 66a 1.89, 66b 1.74, 7a 4.73, 7a 4.73, 7a 4.73, 7 8 5.42, 8 2.91, 9 2.91, 1 7.65, 1 7.75, 1 7.65, 1 7.65, 1 7.65, 1 7.65, 1 7.75, 1 7.65, 1 7.75, 1 7	d, 1.2 s		I		c	
1 5.74, 5.74, 5.74, 5.74, 5.63, 5.63, 5.63, 5.63, 666 1.749, 666 1.749, 666 1.774 8.73, 778 8.82, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91, 2.91	, d, 1.2 s	<sup>13</sup> C	H <sub>1</sub>	<sup>13</sup> C	H	<sup>13</sup> C
3 7.63, 5 5 7.63, 66 66 11.74, 66 7a 4.73 7a 4.73 8 8 4.32, 5.42, 8 9 5.42, 5.42, 5	v.	98.7, d	5.67, d. 1.2	99.1, d	5.56, d, 1.2	99.2, d
4 5 66 6b 11.74, 7a 4.73, 7b 4.73, 8 8 5.42, 5.42, 5.42, 5.42, 5.42, 5.42, 5.42, 5.42, 5.42, 5.42, 5.42, 5.42, 5.42, 5.42, 5.42, 5, 5, 5, 6, 6, 6, 6, 6, 6, 6, 7, 6, 6, 7, 6, 6, 6, 7, 7, 6, 6, 6, 6, 6, 7, 7, 6, 6, 7, 7, 7, 6, 6, 6, 7, 7, 7, 7, 7, 7, 7, 7, 7, 7, 7, 7, 7,	1	154.4, d	7.61, s	154.8, d	7.62, s	154.6, d
5 6a 6b 1.74, 7a 7b 8 8 5.42, 2.91, 2.91,		109.2, s		108.8, s		108.9, s
6a 1.89, 6b 1.74, 7a 4.73, 7b 4.32, 8 5.42, 9 2.91,		64.3, s		64.3, s		64.3, s
6b 1.74, 7a 4.73, 7b 4.73, 8 5.42, 9 5.42, 9 2.91,	m	33.5, t	1.88, m	33.7, t	1.90, m	33.7, t
7a 4.73, 7b 4.32, 8 5.42, 9 2.91,	, bd, 13.4		1.72, bd, 14.0		1.74, bd, 14.1	
7b 4.32, 8 5.42, 9 2.91,	m	65.9, t	4.72, m	66.0, t	4.74, m	65.9, t
8 5.42, 9 2.91,	m		4.33, m		4.33, m	
9 2.91,	m	133.7, d	5.39, m	133.7, d	5.42, m	133.8, d
	, dd, 9.2, 1.3	51.8, d	2.90, dd, 9.2, 1.2	51.9, d	2.90, dd, 9.5, 1.0	52.0, d
10a 5.36,	, dd, 17.0, 2.6	121.3, t	5.34, dd, 17.0, 2.5	121.3, t	5.37, dd, 17.0, 2.3	121.3, t
10b 5.29,	, dd, 9.4, 2.6		5.27, dd, 9.4, 2.5		5.28, dd, 9.6, 2.3	
11		167.9, s		168.1, s		167.9, s
1' 4.76,	, d, 7.9	98.7, d	4.63, d, 8.0	100.0, d	4.64, d, 8.0	100.5, d
2' 3.34,	m	74.4, d	3.28, m	74.4, d	3.26, m	75.2, d
3' 3.62,	. t, 8.7	86.3, d	3.47, t, 9.0	76.5, d	3.44, m	75.7, d
4' 3.30,	m	69.5, d	3.67, t, 9.0	78.3, d	3.77, t, 9.4	79.4, d
5' 3.41,	m	78.6, d	3.37, m	77.5, d	3.40, m	72.4, d
6'a 3.90,	, dd, 12.0, 1.9	62.4, t	3.94, m	61.8, d	3.97, m	67.6, t
6'b 3.69,	, dd, 12.0, 5.4					
1" 4.34,	m	63.4, d	4.41, m	63.8, d	5.78, s	95.0, d
3" 5.50,	S	95.8, d	5.40, s	94.5, d	5.57, s	92.3, d
4″		123.2, s		123.2, d		119.8, s
5"		158.3, s		157.9, s		147.6, s
6"a 2.53, 6"b	, t, 6.3	29.0, t	2.61, m 2.38, m	29.4, t	2.71, m	23.3, t
7″a 4.43,	t, 6.4	67.4, t	4.36, m	66.9, t	4.44, m A 30 m	67.0, t
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9" 26. 9" 1.26.	d. 6.2	20.6. a	2.27, m 1.25, d. 6.2	20.6. a	0.7, H, 1.7	130.7. s
10'		166.5, s		165.2, s	2.01, d, 7.3	14.8, q
11"						165.7, s

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544

## C.-A. Geng et al.



Figure 2. Selected <sup>1</sup>H-<sup>1</sup>H COSY, HMBC, and ROESY correlations of compound 1.

with C-3'. However, only a weak correlation of H-3" with H-9" was detected in the ROESY spectrum, which was insufficient to determine the same orientation of H-3" and Me-9". This problem has been encountered in our previous investigation, namely, swerilactone G possessed a similar partial structure with the fragment 1b, and its relative configuration has been proved by X-ray single-crystal diffraction [16]. Although the Me-9 and H-3 were located at the same side, the ROESY correlation of neither H-3/H-1 nor H-3/H-9 was detected in that the connection of H-C (3)-O-C (1)-C (9) possessed the W conformation. In addition, the coupling constant of H-9" (J = 6.2 Hz) in swerilactoside A was also identical to that of H-9 (J = 6.2 Hz) in swerilactone G. Thus, it was plausible to deduce that fragment 1b possessed a similar configuration to that of 3a in swerilactone G. Consequently, the structure of compound 1 was elucidated to be swerilactoside A, as shown in Figure 1.

Swerilactoside B (2) possessed the same molecular structure of  $C_{25}H_{32}O_{13}$  as that of compound 1. The UV, IR, and NMR spectra of compound 2 were very close to those of compound 1, which suggested that they possessed a similar skeleton. The  ${}^{1}H{-}^{1}H$  COSY and HMBC analyses suggested that compound 2

contained the same partial fragments 2a and 2b as those of compound 1. The HMBC correlations of H-3" with C-4' and H-4' with C-3" and the upfield shift of C-3' from  $\delta_{\rm C}$  86.3 (d) in compound **1** to 76.5 (d) in compound 2, as well as the downfield shift of C-4' from  $\delta_{\rm C}$  69.5 (d) in compound 1 to 78.3 (d) in compound 2, corresponding to the variations of  $\Delta \delta_{\text{H-3}'}$  (-0.15 ppm) and  $\Delta \delta_{\text{H-4'}}$  (+0.37 ppm), proposed that fragment 2b was linked to 2a by C (4')—O—C (3''). Similarly, the correlation of neither H-3"/H-1" nor H-3"/H-9" was detected in the ROESY spectrum (Figure 3), together with the completely consistent coupling constant of H-9'' (J = 6.2 Hz) with that in swerilactoside B and swerilactone G [17], which indicated that fragment 2b adopted the same relative configuration as that of 1b. Thus, the structure of compound 2 was elucidated to be swerilactoside B, as shown in Figure 1.

Swerilactoside C (**3**) had a molecular formula of  $C_{26}H_{30}O_{13}$  by a quasi-molecular ion peak at m/z 585.1367 [M + Cl]<sup>-</sup> in the negative HR-ESI-MS. The IR spectrum suggested the presence of OH (3436 cm<sup>-1</sup>), C=O (1703 cm<sup>-1</sup>), double bond (1620 cm<sup>-1</sup>), and glycosyl group (1084, 1057, and 1032 cm<sup>-1</sup>).

The <sup>1</sup>H and <sup>13</sup>C NMR (DEPT) spectra exhibited 26 carbon resonances due to 7



Figure 3. Selected <sup>1</sup>H-<sup>1</sup>H COSY, HMBC, and ROESY correlations of compound 2.

quaternary carbons, 12 methines, 6 methylenes, and 1 methyl group. The NMR spectral data of compound 3 were similar to those of compound 2, except for the presence of additional tri-substituted double bond [ $\delta_{\rm C}$ 138.4 (d, C-8") and 130.7 (s, C-9")] and the absence of one methylene [ $\delta_{\rm C}$  37.4 (t, C-8")] observed in compound 3, together with the obvious downfield shift of C-1" [from  $\delta_{\rm C}$ 63.8 (d) in compound **2** to  $\delta_{\rm C}$  95.0 (d) in compound 3]. In addition, the chemical shift variation of C-2' ( $\Delta \delta = +0.8 \text{ ppm}$ ), C-3'  $(\Delta \delta = -0.8 \text{ ppm}), \text{ C-5}' (\Delta \delta = -5.1 \text{ ppm}),$ C-6' ( $\Delta \delta = +5.8$  ppm), and C-10'' [ $\Delta \delta =$ -5.8 ppm (corresponding to C-9" in compound 2)] was observed in Table 1. In addition to the swertiamarin fragment (3a), the other partial structure 3b was constructed based on the <sup>1</sup>H-<sup>1</sup>H COSY correlations of H-6" with H-7", and H-8" with H-10", and the HMBC correlations of H-7" with C-5" and C-11", H-6" with C-4" and C-9", H-8" with C-5" and C-1", and H-3" with C-1", C-5" and C-11" (Figure 4). The glycosidic linkage between C-1" and C-4' was deduced by HMBC correlations of H-1"/C-4' and H-4'/C-1". Similarly, the connection of C-3" and C-6' by a glycosidic bond was detected by the HMBC correlations of H-3" with C-6' and H-6' with C-3". The correlations of H-1"/H-4' and H-3"/H-6' in the ROESY spectrum suggested the  $\beta$ -orientation of H-1" and the  $\alpha$ -orientation of H-3". The Z-configuration of the double bond between C-8" and C-9" was deducted based on the ROESY correlations of H-8" with H-6" and H-10" with H-1". Thus, the structure of compound **3** was deduced to be swerilactoside C, as shown in Figure 1.

Swerilactosides A–C were three unusual secoiridoid glycoside dimers obtained from the traditional Chinese herb *S. mileensis*, which further enriched the skeleton type of secoiridoid glycosides.

### 3. Experimental

## 3.1 General experimental procedures

Optical rotations were determined on a JASCO model 1020 polarimeter (Horiba, Tokyo, Japan). UV spectra were measured on a Shimadzu UV-2401A spectrophotometer (Shimadzu, Kyoto, Japan). IR (KBr) spectra were recorded on a Bio-Rad FTS-135 spectrometer (Bio-Rad, Hercules, CA, USA). 1D and 2D NMR spectra were recorded on Bruker AM-400 NMR or DRX-500 spectrometers (Bruker, Bremerhaven, Germany) with TMS as an internal standard. MS spectra were run on a VG Auto Spec-3000 spectrometer (VG, Manchester,



Figure 4. Selected <sup>1</sup>H-<sup>1</sup>H COSY, HMBC, and ROESY correlations of compound 3.

England). Silica gel (200–300 mesh) for column chromatography was obtained from Qingdao Makall Chemical Company, Qingdao, China. HPLC (Waters Alliance 2695), equipped with a photodiode array detector (Waters 2996) and a Waters 600 pump, was purchased from Waters Co. Ltd, Milford, MA, USA. Sephadex LH-20 (20–150  $\mu$ m) was purchased from Pharmacia Fine Chemical Co. Ltd, Uppsala, Sweden.

## 3.2 Plant material

The whole plant of *S. mileensis* was collected in Mile County, Yunnan Province, China, on 6 November 2008, and was identified as *S. mileensis* T. N. Ho et W. L. Shi by Prof. Dr. Li-Gong Lei, Kunming Institute of Botany, Chinese Academy of Sciences. A voucher specimen (No. 2008-11-01) has been deposited in the Laboratory of Antivirus and Natural Medicinal Chemistry, Kunming Institute of Botany.

#### 3.3 Extraction and isolation

The air-dried whole plant (5.0 kg) of *S. mileensis* was powdered and extracted with 90% and 50% EtOH under reflux successively (each time 2 h, 15.0 liters  $\times$  2 times). The combined extracts were concentrated under reduced pressure to give a residue (1.3 kg). The residue was suspended in water and extracted with

petroleum ether (1.0 liters  $\times$  2), ethyl acetate (1.0 liters  $\times$  3), and *n*-butanol (1.0 liters  $\times$  3) successively. The ethyl acetate part (170.5 g) was chromatographed on a silica gel column (2.0 kg,  $11.0 \times 50.0$  cm) eluted with CHCl<sub>3</sub>-MeOH (from 100:0 to 0:100, v/v) to furnish 10 fractions A-J. Fraction B (8.5 g) was chromatographed on a silica gel column  $(100.0 \text{ g}, 3.0 \times 30.0 \text{ cm})$  with a gradient elution of CHCl<sub>3</sub>–Me<sub>2</sub>O (90:1  $\rightarrow$  50:50) to supply four fractions B1-B4. Fraction B4 (3.0 g) was performed on a silica gel column  $(30.0 \text{ g}, 1.7 \times 25.0 \text{ cm})$  eluted with CHCl<sub>3</sub>-MeOH (90:1  $\rightarrow$  80:20) to obtain three subfractions B4-1 to B4-3. Subfraction B4-1 (100.0 mg) was dissolved in MeOH and purified with a semi-preparative HPLC apparatus, using a Waters XTerra Prep RP-18 column (7.8  $\times$  300 mm, 10  $\mu$ m), eluted with MeOH-H<sub>2</sub>O (35:65, flow rate = 4.5ml/min), detected at 254 nm, to obtain compound **1** (80.0 mg, Rt = 18.0 min). Subfraction B4-2 (500.0 mg) was subjected to a silica gel column (30.0 g,  $1.7 \times 25.0$  cm) eluted with CHCl3-Me2CO (80:20), and then further purified with HPLC (the conditions were similar to compound 1) to supply compound 2 (30.0 mg, Rt =13.0 min). Subfraction B4-3 (300.0 mg) was first loaded on a silica gel column  $(30.0 \text{ g}, 1.7 \times 25.0 \text{ cm})$  and eluted with CHCl<sub>3</sub>-Me<sub>2</sub>O (80:20), and then purified with a Sephadex LH-20 column (50.0 g,  $1.4 \times 145.0$  cm, MeOH) to give compound **3** (100.0 mg).

#### 3.3.1 Swerilactoside A (1)

A white powder;  $[\alpha]_D^{19.8} - 94.04$  (c = 0.68, MeOH); UV (MeOH)  $\lambda_{max}$  (nm) (log  $\varepsilon$ ): 231 (4.16); IR (KBr)  $\nu_{max}$  (cm<sup>-1</sup>): 3423, 1698, 1620, 1473, 1419, 1280, 1269, 1082, 1027, 1005, 787; <sup>1</sup>H and <sup>13</sup>C NMR spectral data see Table 1; ESI-MS (+) *m*/*z*: 563 [M + Na]<sup>+</sup>; HR-ESI-MS (+) *m*/*z*: 563.1728 [M + Na]<sup>+</sup> (calcd for C<sub>25</sub>H<sub>32</sub>O<sub>13</sub>Na, 563.1740).

### 3.3.2 Swerilactoside B (2)

A white powder;  $[\alpha]_D^{19.8} - 127.69$  (c = 0.20, MeOH); UV (MeOH)  $\lambda_{max}$  (nm) (log  $\varepsilon$ ): 231 (4.12); IR (KBr)  $\nu_{max}$  (cm<sup>-1</sup>): 3429, 1705, 1620, 1472, 1416, 1326, 1280, 1207, 1154, 1079, 1028, 948, 929, 758; <sup>1</sup>H and <sup>13</sup>C NMR spectral data see Table 1; ESI-MS (–) m/z: 575 [M + Cl]<sup>-</sup>; HR-ESI-MS (–) m/z: 575.1517 [M + Cl]<sup>-</sup> (calcd for C<sub>25</sub>H<sub>32</sub>O<sub>13</sub>Cl, 575.1531).

## 3.3.3 Swerilactoside C (3)

A white powder;  $[\alpha]_{D}^{20.0} - 67.11$  (c = 0.14, MeOH); UV (MeOH)  $\lambda_{max}$  (nm) (log  $\varepsilon$ ): 269 (4.14), 240 (4.14); IR (KBr)  $\nu_{max}$  (cm<sup>-1</sup>): 3436, 1703, 1620, 1472, 1434, 1408, 1273, 1246, 1208, 1159, 1126, 1084, 1057, 1032, 1013, 961, 930, 903, 846, 760; <sup>1</sup>H and <sup>13</sup>C NMR spectral data see Table 1; ESI-MS (-) m/z: 585 [M + Cl]<sup>-</sup>; HR-ESI-MS (-) m/z: 585.1367 [M + Cl]<sup>-</sup> (calcd for C<sub>26</sub>H<sub>30</sub>O<sub>13</sub>Cl, 585.1374).

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#### References

- T.N. Ho, *Flora of China* (Science Press, Beijing, 1988), Vol. 62.
- [2] W.X. Yang, L. Zhou, H.L. Geng, and B.F. Qin, Acta Bot. Real. Occident. Sin. 23, 2235 (2003).
- [3] S.R. Jensen and J. Schripsema, in Gentianaceae-Systematic and Natural History, edited by L. Struwe and V. Albert (Cambridge University Press, Cambridge, 2002), Chap. 6, p. 573.
- [4] F.Y. Fu and N.C. Sun, Acta Pharm. Sin. 6, 198 (1958).
- [5] X.T. Liang, D.Q. Yu, and F.Y. Fu, Acta Pharm. Sin. 11, 412 (1964).
- [6] Z. Xue and X.T. Liang, *Chin. Sci. Bull.* 19, 378 (1974).
- [7] T.R. Govindachari, S.S. Sathe, and N. Viswanathan, *Indian J. Chem.* 4, 201 (1966).
- [8] P.T. Li, A.J. Leeuweberg, and D.J. Middleton, *Flora China* 16, 115 (1995).
- [9] L. Gao, Yunnan J. Trad. Chin. Med. Mater. Med. 27, 65 (2006).
- [10] Y. Du and H.D. Li, *Chin. Pharm.* 17, 304 (2006).
- [11] Q.S. Liang and X.Y. Gao, *Zhong Cao Yao Tong Xun* 9, 1 (1979).
- [12] G.M. Du and G.Y. Li, Yunnan Zhong Yi Za Zhi 3, 35 (1981).
- [13] Contagious Department, 59th Hospital of the Chinese PLA, *New Chin. Med.* 4, 202 (1975).
- [14] X.S. Li, Z.Y. Jiang, F.S. Wang, Y.B. Ma, X.M. Zhang, and J.J. Chen, *China J. Chin. Mater. Med.* 33, 2790 (2008).
- [15] C.A. Geng, Z.Y. Jiang, Y.B. Ma, J. Luo, X.M. Zhang, H.L. Wang, Y. Shen, A.X. Zuo, J. Zhou, and J.J. Chen, *Org. Lett.* **11**, 4120 (2009).
- [16] C.A. Geng, X.M. Zhang, Y. Shen, A.X. Zuo, J.F. Liu, Y.B. Ma, J. Luo, J. Zhou, Z.Y. Jiang, and J.J. Chen, *Org. Lett.* **11**, 4834 (2009).
- [17] C.A. Geng, X.M. Zhang, Y.B. Ma, Z.Y. Jiang, J. Luo, J. Zhou, H.L. Wang, and J.J. Chen, *Tetrahedron Lett.* **51**, 2483 (2010).
- [18] S. Rodriguez, A. Marston, J.L. Wolfender, and K. Hostettmann, *Curr. Org. Chem.* 2, 627 (1998).
- [19] C.H. Duan, B.J. Shi, L.H. Wu, G.X. Chou, and Z.T. Wang, *Chin. J. Nat. Med.* 5, 417 (2007).